FLURY Virginie (2016): Biogeochemical Study of Uranium Accumulation in an Alpine HISTOSOL (Davos, Switzerland)

<u>Abstract</u>

In 2005, some hotspots containing high concentrations of uranium (more than 4000 ppm) were discovered in the Dischma Valley (Davos, Switzerland) (FROIDEVAUX et al., 2006; REGENSPURG et al., 2010). To better understand the mobility of uranium and its accumulation mechanism, a core of a uranium-rich HISTOSOL was transferred in 2011 to a large laboratory lysimeter maintained under semi-natural conditions. Previous solute flux on this core revealed the activity of anaerobic sulfate- and iron-reducing bacteria (REGENSPURG et al., 2010). In the current study, an amendment of depleted U6+ (U-238/U-234 = 2.38), Fe3+ and SO42- was made and the species in pore water were studied over a 4-month period, in order to test the hypothesis that uranium co-precipitates with mackinawite (FeS(s)), a byproduct mineral of sulfate- and iron-reducing bacteria. U6+ was analysed by α Spectrometry. This experiment failed to confirm the uranium coprecipitation with mackinawite. Indeed, SO42- was measured in the pore water indicating no reduction of SO42-, whereas the reduction of NO3- was observed. These observations indicate a possible shift in the bacteria population from the previous amendment to the current one. After the 4-month column experiment, a sub-core was taken. The measurements of the soil samples by y Spectrometry revealed the precipitation of fresh uranium in the upper layer of the soil. Some significant correlations were observed between uranium, calcium and sulfur after analysing the soil samples by X-ray Fluorescence Spectrometry. Also, LEZAMA-PACHECO et al. (2015) have shown that Ca2+ was present at the surface of uraninite nanoparticules and might decrease the dissolution rate of uraninite. After several years of multiple amendments, the Dischma's core still contains high uranium content and is not significantly disturbed. The study of LEZAMA-PACHECO et al. (2015) might explain partially the accumulation of U in this anoxic environment.